Influence of Pulse Energies on the Structure SnO$_2$ Films Plasma

Alansari RA

Department of Physics, College of Science, University of Baghdad, Baghdad, Iraq

Review Article

ABSTRACT

SnO$_2$ films were prepared by pluse laser deposition (PLD) technique. The Pulsed Nd:YAG laser was used for prepared SnO$_2$ thin films under O$_2$ gas environment with varying pulse energies. X-ray diffraction patterns and intensity curves for the SnO$_2$ films on glass substrates obtained by PLD technique show that the SnO$_2$ polycrystalline films with all energies. The optical properties of as-grown film such as optical transmittance spectrum, refractive index and energy gap have been measured experimentally and the effects of laser pulse energy on it were studied

INTRODUCTION

Tin oxide films (SnO$_2$) have been studied with a focus on application to sensors, transparent electrodes in displays, heat mirrors and transparent conducting oxide (TCO) coatings for solar cells [1-5]. SnO$_2$ films have been prepared by several deposition techniques, such as pulse laser deposition (PLD), photo-MOCVD, reactive evaporation, spray paralysis, sol-gel process, and dc/rf sputtering [6-8]. PLD is the plasma produced by the interaction of high-energy laser pulses with matter in any state of aggregation [9,10]. Laser induced plasmas of metals and alloys are of great interest since they have different attractive and important applications, e.g. material processing, thin film deposition, the synthesis of nanoparticles, the elemental analysis of multi component materials, precision machining, and laser induced breakdown spectroscopy (LIBS), surgery, and laser micro-probe mass spectroscopy [11-13].

Typical Experimental Set-Ups

PLD experiment was carried out under vacuum pressure (6 x 10^{-2} mbar by using Varian DS219 Rotary pump). The beam of Nd:YAG laser with second harmonic frequency (λ=532 nm, 10 ns, 6 Hz) was focused onto target. The target of the deposition was SnO$_2$ bulk with purity 99.999%, shaped liked disc with a diameter of 1 cm. the target was kept onto rotating holder (speed 4 rev/min) to prevent fast drilling. The substrate distance from the target was fixed to 2 cm,

The PLD experiment was performed at room temperature and the as-grown samples were not annealed after deposition. PLD setup scheme has been shown in **Figure 1**. The crystalline structure was examined using X-ray diffraction (XRD). Optical properties (UV/VIS absorption spectrum) of the SnO$_2$ films were performed using. (UV-Visible spectrometer). The laser pulse energy was varied from (400-800) mJ with increment.

Figure 1. Pulsed laser deposition (PLD) system.

RESULTS AND DISCUSSIONS

X-Ray Diffraction Spectra

The XRD patterns of the SnO$_2$ thin film is given in **Figure 1**. **Figures 2a-2c** shows the X-ray diffraction patterns for undoped SnO$_2$ films grown on glass substrates We can be seen that the degree of crystalline and grain size variation with laser energy this can be interoperated in term the improvement the crystal stricture of these films with increasing the laser energy, it seems that all the films are polycrystalline [11]. The grain size (D) of the material; which plays an important role in the material properties, can be estimated easily from the X-ray spectrum by means of full width at half maximum (FWHM) method that is often calculated by Scherrer's relation [9],

$$D = \frac{k\lambda}{\beta.\cos\theta}$$

Where λ is the wavelength of X-ray used (1.54 A), βis the full width half maximum (FWHM) of the peak and θ is the glancing angle. The calculated crystalline size (D) of Tin oxide is tabulated in **Table 1**. The FWHM and the grain size of the samples are shown in **Table 1**.
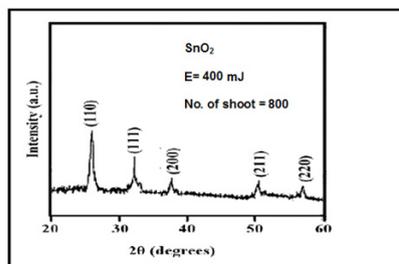

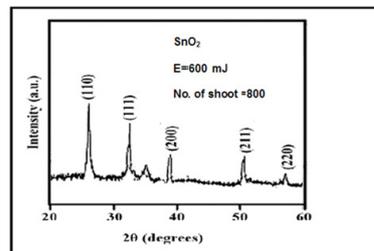**Figure 2a:** XRD of SnO$_2$ thin film with Number of shoot=400 and different energies.

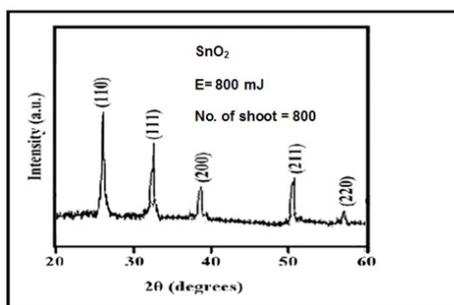**Figure 2b:** XRD of SnO$_2$ thin film with Number of shoot=600 and different energies.

Figure 2c: XRD of SnO_2 thin film with Number of shoot=800 and different energies.

Table 1: Thin film and different energies.

Energies	2θ (Deg.)	FWHM (Deg.)	d_{hkl} Exp. (Å)	G.S (nm)	hkl
400	26.22	0.219	3.3961	371	-110
	32.47	0.247	2.7551	338	-111
	38.27	0.243	2.3499	301	-200
	50.78	0.25	1.7965	278	-211
	58.01	0.257	1.588	246	-220
600	26.27	0.244	3.3879	321	-110
	32.75	0.219	2.7323	316	-111
	38.89	0.205	2.31	298	-200
	50.31	0.149	1.8122	286	-211
	58.27	0.114	1.5822	269	-220
800	26.4	0.19	3.3733	327	-110
	33.1	0.183	2.704	319	-111
	83.92	0.162	2.31	267	-200
	50.49	0.144	1.8	298	-211
	58.6	0.165	1.574	243	-220

UV-Visible Spectroscopy

The optical properties of SnO_2 have been determined by using spectrophotometer in the wavelength range (200-1100) nm.

SnO_2 thin films were successfully deposited onto glass substrate and the films were very transparent.

Transmittance spectra recorded for SnO_2 films as a function of wavelength range (200-1100 nm) at different energies 400, 600, and 800 mJ shown in **Figure 3**.

This **Figure 4** show that the transmittance decreases with increasing of energies due to increasing of thicknesses.
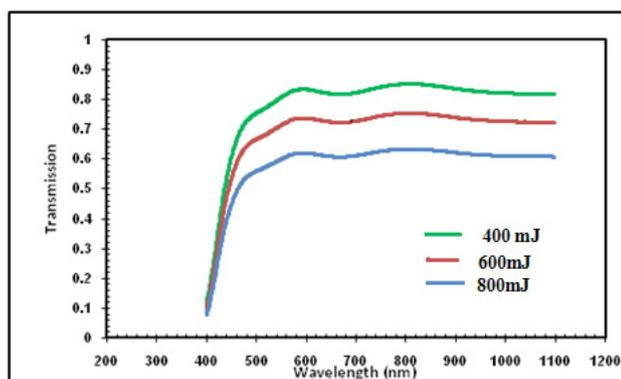

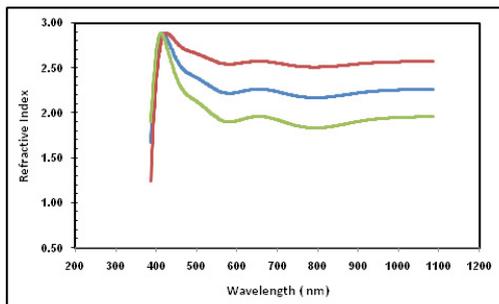**Figure 3:** Shows the Transition of SnO_2 thin films as a function of wavelength.

Figure 4: Shows the Refractive index of SnO_2 thin films as a function of wavelength.
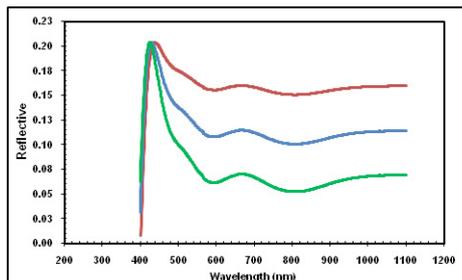
Figure 5: Shows the Reflective of SnO_2 thin films as a function of wavelength.

The absorption coefficient (α) is calculated using the equation, [12]

$$\alpha = \ln (1/T)/d \qquad (1)$$

Where T is transmittance and d is film thickness (**Figure 5)**. The absorption coefficient (α) and the incident photon energy (hν) are related by the following equation [13]:

$$(\alpha h\nu)2 = A (h\nu - Eg) \qquad (2)$$

The typical plots of $(\alpha h\nu)^2$ versus hν for SnO_2 thin films with (400, 600, and 800) mJ energies deposited on glass substrate is shown in **Figure 6**. It is observed that increase in energies of laser lead to increase in optical band gap from 2.756 eV to 2.9 eV. This may be associated with variation the crystal structure with laser energies.
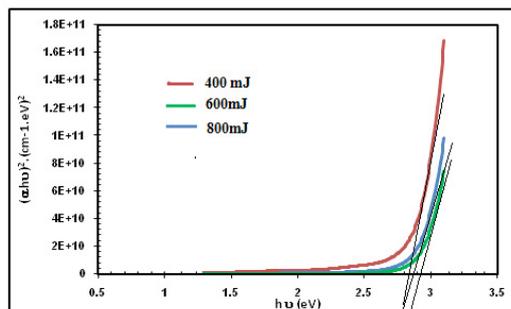
Figure 6: Shows the direct electronic transition for nanostructure Sno_2 thin films.

CONCLUSIONS

In this work we have reported the influence of energies of laser in the structural and optical characteristics of SnO_2 thin films.

We describe bellow summarization of our work:

1. The deposition films are having the polycrystalline structure of Tin oxide with cubic structure.

2. The intensity of x-ray diffraction was proportional with laser energies

3. The optical properties was proportional with laser energies.

4. The band gap increase with increasing of laser energies.

REFERENCES

1.	Stambolova K, et al. Lanthanum doped SnO2 and ZnO thin films sensitive to ethanol and humidity. Mater Ce Phys. 2000;63:104.

2.   Chen M, et al. ZAO: an attractive potential substitute for ITO in flat display panels. Mater. Sci. Eng. B 2001;85:212.

3.   Rye YR, et al. Synthesis of p-type ZnO films. Journal of Crystal Growth. 2000;216:330.

4.   Martinez MA, et al. Deposition of transparent and conductive Al-doped ZnO thin films for photovoltaic solar cells. Solar Energy Materials and Solar Cells. 1997;45:75.

5.   Reddy KTR, et al. Effect of gallium incorporation on the physical properties of ZnO films grown by spray pyrolysis. J Cryst Growth. 2000;210:516.

6.   Song J, et al. Electrical and optical properties of ZnO thin films prepared by the pyrosol method. J Korean Phys Soc. 1996;29:219.

7.   Lee JH, et al. Electrical and optical properties of ZnO transparent conducting films by the sol-gel method. J Cryst Growth. 2003;247:119.

8.   Smith HM and Turner AF. Vacuum deposited thin films using a ruby laser. Appl Opt. 1965;4:147.

9.   Dijkkamp D, et al. Preparation of Y-Ba-Cu oxide superconductor thin films using pulsed laser evaporation from high T_c bulk material. Appl Phys Lett. 1987;51:619.

10.   Chrisey DB and Hubler GK. Pulsed laser deposition of thin films. Wiley, New York, 1994.

11.   Scarfone C, et al. Pulsed Laser Deposition (PLD) -- A Versatile Thin Film Technique. Mat Res Soc Symp Proc. 1991;191:183.

12.   Fahler S, et al. Origin and avoidance of droplets during laser ablation of metals. Appl Surf Sci. 1997;109:433.

13.   Burukhin DS, et al. Hydrothermal Synthesis of Nanocrystalline SnO2 for Gas Sensors. Inorganic Materials. 2003;39:1158-1162.