

Characterization of Biodiesel Obtained From Pure Soybean Oil and Its Various Blends with Petro-Diesel

A.K. Goswami¹, G. A. Usmani²

¹ Associate Professor, Department of Chemical Engineering, University Institute of Chemical Technology, North Maharashtra University, Jalgaon, Maharashtra, India.

² Head & Professor, Department of Oil Technology, University Institute of Chemical Technology, North Maharashtra University, Jalgaon, Maharashtra, India.

ABSTRACT: The current study is focused on alternative fuel for diesel in the form of soybean oil methyl ester and their different blends with petroleum diesel. The soybean oil is important oil from economic point of view for the biodiesel production and in the country like India this oil is mainly used for edible purpose. The Transesterification process was used to produce the methyl ester from soybean oil. The methyl-ester prepared from soybean oil was blended with diesel fuel such as 10% biodiesel plus 90% Petro-diesel, 20% biodiesel plus 80% Petro-diesel, 30% biodiesel plus 70% Petro-diesel, 40% biodiesel plus 60% Petro-diesel, 60% biodiesel plus 40% Petro-diesel and 80% biodiesel plus 20% Petro-diesel. In the present study, the petroleum analysis of soybean oil, biodiesel (soybeanoil methylester) and its different blends with diesel were done and studied. Properties such as fire and flash point, pour point, aniline point, diesel index, specific gravity and cetane number were determined and also distillation characteristics consisting of initial boiling point and final boiling point were also be studied.

KEYWORDS: Aniline Point, Biodiesel, Diesel Index, Kinematic Viscosity, Soybean Oil, Transesterification.

I. INTRODUCTION

Energy and Fuel are the key aspects for mechanical, modern, social and temperate advancement of any country. Reduction in fossil reserves and higher costing are bringing about a developing requirement for substituting the ordinary fossil fuel with powers inferred from vegetable source otherwise called biodiesel [1 – 4]. Biodiesel is an alternative fuel which is derived from biomass. Biodiesel is obtained from the transesterification of fats and oils and this biofuel has a comparable properties to that of diesel fuel acquired from petroleum source [5 – 7]. One of the fundamental benefit of biodiesel is that it is biodegradable in nature and might be utilized without adjusting or modification in the current motors or engine and likewise delivers minimum pollutant gas in the exhaust gas discharges [8 – 12]. An additional essential thing of biodiesel is that it is miscible with diesel and might be very effectively mixed [13]. The requirement of any fuel that it must be actually and naturally achievable, satisfactory and monetarily aggressive to give the fuel performance [14]. The transesterification process yields to the biodiesel production by large alludes to a catalyzed synthetic reaction of a vegetable oil and liquor to results in an unsaturated fat alkyl ester and glycerol as byproduct [15]. The unsaturated fat alkyl ester or any unsaturated fat methyl or propyl ester (otherwise known as FAME) is the biodiesel [16, 17]. This is a reversible reaction, in which excess amount of liquor is utilized to give a maximum yield. Most of the reactions, methanol is the most favorable liquor due to having low comparative cost along with suitable physical and chemical properties.

Soybean oil has a good nutritious quality. It holds around 20% oil and 40% protein (as against 7.0 % in rice, 12 % in wheat, 10% in maize and 20-25% in different beats). Soybean protein is rich in significant amino acid lysine (5%). It holds an excellent measure of minerals, salts and vitamins (thiamine and riboflavin) and its grains in growing stage contains Vitamin C, Vitamin A in sufficient quantity. Soybean oil is the less expensive and possessing best quality

International Journal of Innovative Research in Science, Engineering and Technology

(An ISO 3297: 2007 Certified Organization)

Vol. 3, Issue 9, September 2014

proteins and fats. Soybean oil is the second most widely used vegetable oil after palm oil. Palm and soybean oils together constitute approximately 68% of global edible oil. Only Soybean oil constituting around 22.85%. In the recent decade, volume of Soybean oil traded has developed at the rate of 4.05% which indicates around 25% of the global aggregate oils and fats production. Brazil, China, Argentina, and India alongside USA are the significant contributors of soybean oil for the development. India is the World's biggest shipper of edible oils in the World, of an aggregate 5.0-5.5 million tons of vegetable oils of foreign made, 1.3-1.5 million tons is Soybean oil. Unrefined and Refined soybean oils are permitted to be imported into India at an import obligation of 45% and 50.4% separately. Indian consumable oil business environment is most value touchy in nature. The climatic necessities for soybean practically the similar to maize. A temperature of 26.5 to 30 °C seems, by all accounts, to be the ideal for the majority of the assortments. Soil temperatures of 15.5 °C or above favorable for germination and energetic seedling development. The base temperature for compelling development is around 10°C. A lower temperature than 10°C, has a tendency to delaying of the blooming. [Ministry of Agriculture, India].

II. METHADODOLOGY

The soybean oil has been utilized for preparation of biodiesel (alkyl ester of soybean oil). In a measuring cylinder some pellets of potassium hydroxide is taken and methanol is made to react as demonstrated by stoichiometric proportion. The catalyst pellets were well shaken and blended with the methanol. At the opening of the flask, the stirrer is properly fitted. The oil was heated to 60 °C to remove moisture content and then mixed with potassium methoxide. The mixture heated continuously at 65 °C. After two hours, the response of reaction mixture changed to a turbid orange tan color. The sweet smell of the reaction mixture was observed which representing the finishing stage of reaction. The process was continued for another ten to fifteen minutes. The two layers were formed, one is ester and another one is glycerol. In the ester phase, certain amount of unreacted methanol, removed by passing 100 to 150 ml of warm water. The glycerol, methanol was separated while the ester phase hold little measure of water. Water is removed by passing ester through Ca(OH)₂ pellet. The methyl ester of pure soybean oil was obtained. This soybean oil methyl ester was blended with diesel fuel in various proportions and the various analysis such as acid value, specific gravity, density, kinematic viscosity, aniline point, cloud point, pour point, flash point and cetane number was studied and results were compared with the standard values of specification of the diesel fuel.

III. STUDY OF VARIOUS PROPERTIES OF BIODIESEL

1. Kinematic viscosity determination:

The acceptable range of kinematic viscosity for diesel fuel at 40°C to be 2.5 to 7.5 cS as prescribed by BIS standards. For pure soybean oil, the kinematic viscosity was found, 7.54 cS which is slightly above the required range while the pure diesel has a kinematic viscosity of 2.75 cS and pure soybean oil biodiesel has a kinematic viscosity of 6.22 cS, which is comparatively higher but within the range. The viscosity of 10%, 20%, 30%, 40%, 50%, 60%, 80% blends of methyl ester of pure soybean with diesel are 3.02, 3.35, 3.61, 4.12, 4.85, & 5.71 centistokes respectively. The viscosity of biodiesel blends from 10 to 30% were between 3 and 4 cS. This result represents these blends are highly favorable as per the proper ignition due to lower values of viscosity. These lower blends are capable to fulfill the demand of alternative fuel by saving diesel and effectively utilized for better performance of diesel engine.

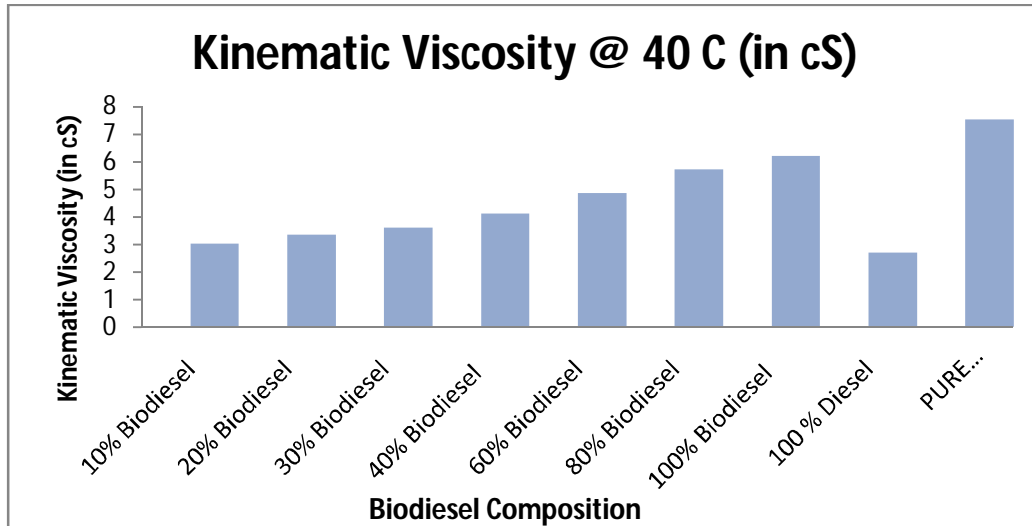


Fig 1: shows change in kinematic viscosity with changes in biodiesel composition

2. Diesel Index:

Diesel index is an indicator of the good ignition quality of diesel fuel. The diesel index for diesel has to be 45 min.as per BIS standards.

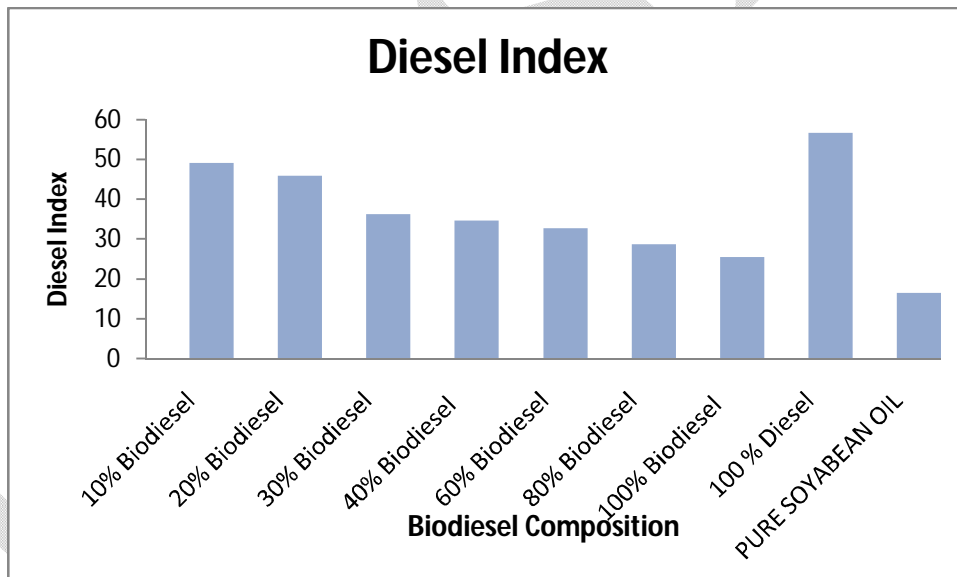


Fig 2: Shows changes in diesel index with changes in biodiesel index

The diesel index of pure soybean oil is 16.6 which is low whereas diesel has a diesel index of 56.63 minutes. While the soybean oil methyl ester, diesel index was found 25.5 which was much lower than diesel. Thus use of pure soybean methyl ester as a fuel will result in a very poor ignition quality. Hence, pure soybean oil methyl ester cannot be used as a fuel alone and blending of biodiesel is preferred with petro diesel. Blend of 10% and 20% methyl ester of pure soybean oil have Diesel Index of 49.17 and 45.86 respectively, which are above the BIS value for diesel index and thus these blends will have superior ignition quality. Other blends of 30%, 40%, 60% and 80% have a lower diesel index as a result causes a poor ignition. As the percentage of blends increase, decrease in the Diesel Index values was observed. So the 10 and 20% blends are most favorable.

3. Aniline Point:

When the fuel is highly paraffinic it results in high Diesel Index, hence resulting in high aniline point, which is an indication of superior ignition properties. Evaluation of aniline point is important in determining the diesel index. The various aniline points for various blends have been shown in the graph.

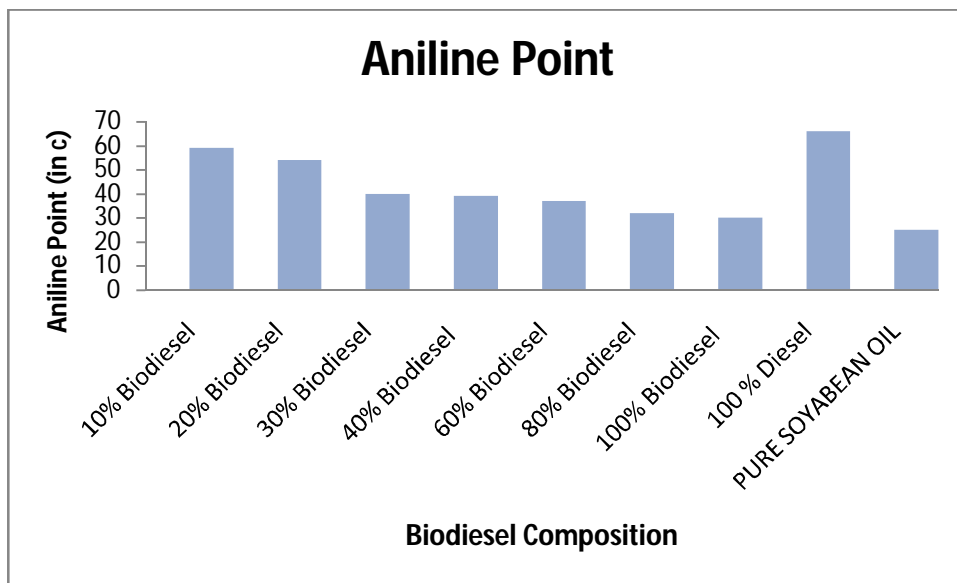


Fig 3: Shows change in Aniline point with changes in biodiesel composition

High speed diesel i.e. 100% diesel has an aniline point of 66 °C whereas 100% biodiesel has a much lower aniline point of 25 °C. The blends of pure soybean methyl ester and high speed diesel have a comparatively higher aniline point when compared to high speed diesel but still less than pure biodiesel obtained from the pure soybean oil.

4. Pour Point:

Pour point is the lowest temperature at which it has utility in certain application. BIS Specification for max pour point is +6 °C and the market purchased diesel has a pour point of +5 °C and it thus meets the requirements Pour point of pure soybean oil is -6°C which is far below the requirements thus using pure soybean oil as a fuel is of no problem from pour point aspects. Pour point of methyl ester of pure soybean oil (100%) and 10%, 20%, 30%, 40%, 60% and 80% blends of methyl ester of pure soybean oil with diesel are 2°C, 4°C, 3°C, 3°C, 2°C and 2°C respectively and are within the specified range given BIS. Thus increasing amount of biodiesel decreases the pour point of biodiesel. Blending above 20% biodiesel is not permissible because diesel index does not satisfy the requirement which is much important than pour point and kinematic viscosity.

5. ASTM Distillation Properties:

As per BIS specification of diesel 90% distillate temperature should be below 366 °C. ASTM distillation of soybean oil is not possible because of cracking that takes place at higher temperature. Thus considering ASTM characteristics, kinematic viscosity and diesel index as a property we cannot use pure oil in the diesel engine as per the standard properties specification. The graph for ASTM Distillation for Initial Boiling Point, Final Boiling Point and for 90% recovery is as shown below:

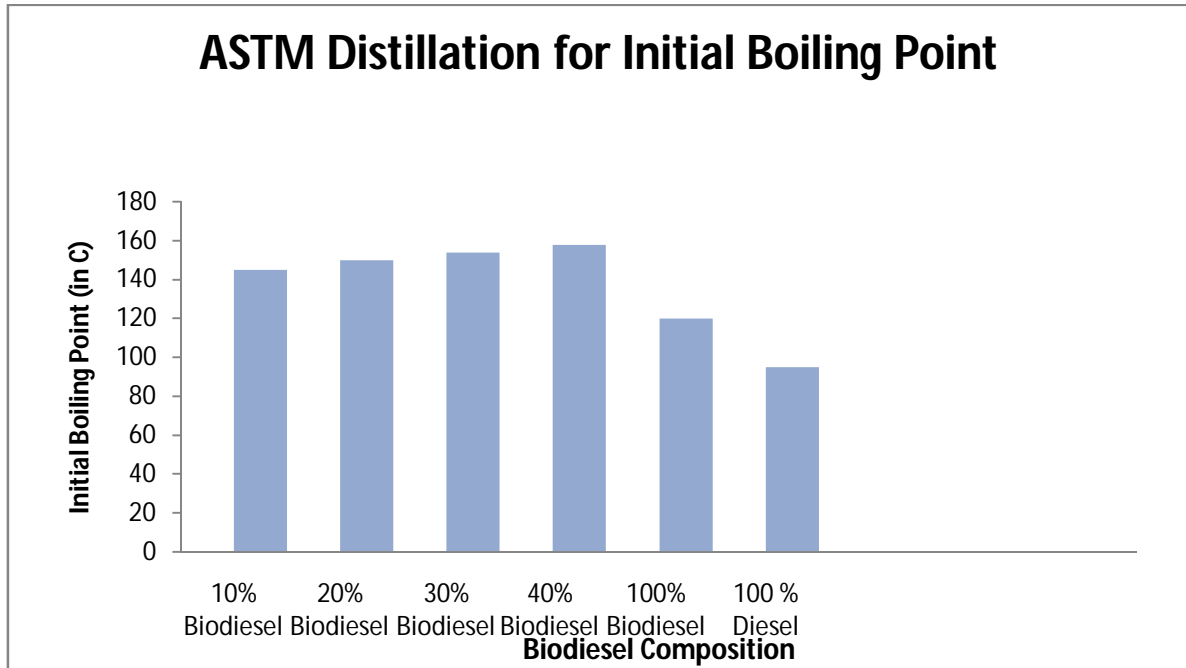


Fig 4: Shows various Initial Boiling points for various composition of biodiesel

For Initial Boiling Point, temperature of 95 °C is obtained for 100% diesel whereas temperature of 120 °C is obtained for 100% biodiesel. ASTM Distillation for Final Boiling Point is shown by the graph below:

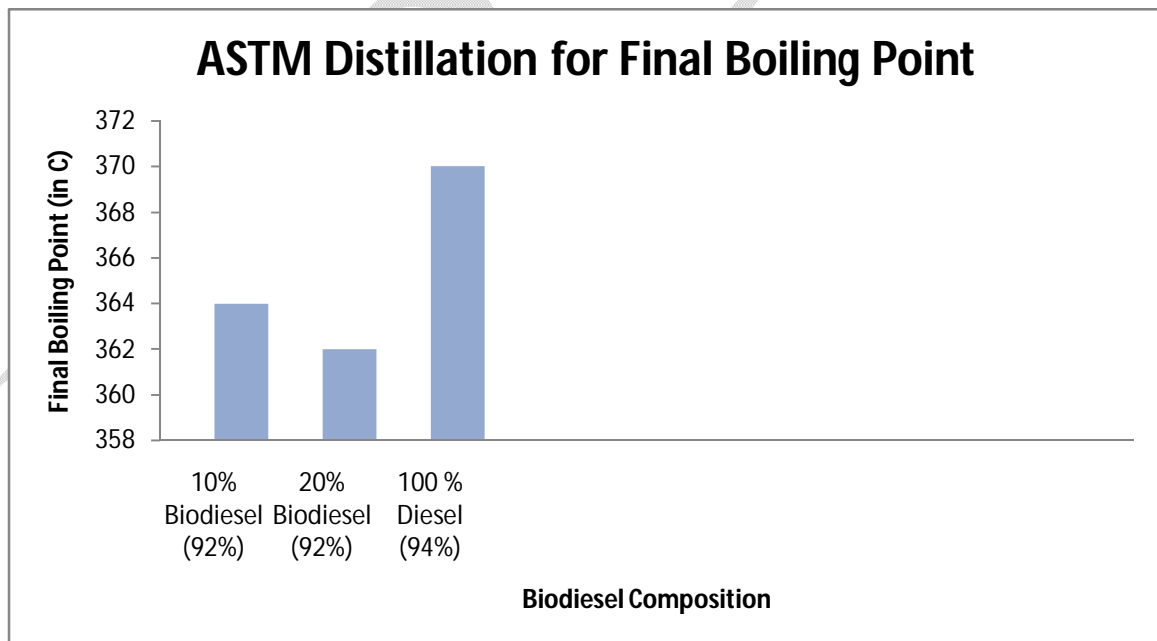


Fig 5: Shows various Final Boiling points for various composition of biodiesel

International Journal of Innovative Research in Science, Engineering and Technology

(An ISO 3297: 2007 Certified Organization)

Vol. 3, Issue 9, September 2014

From the graph we can categorically conclude that for both 10% biodiesel and 20% biodiesel similar distillate amount of 92% is obtained for almost similar temperatures. For blends above 20%, cracking take place and FBP is not obtained.

6. Comparison of Miscellaneous properties:

Properties	100 % Diesel	10% Biodiesel	20% Biodiesel	30% Biodiesel	40% Biodiesel	60% Biodiesel	80% Biodiesel	100% Biodiesel	Soybean Oil
Specific Gravity at 15 °C	0.837	0.8472	0.849	0.8506	0.855	0.859	0.865	0.878	0.924
API Gravity	37.556	35.58	35.5	34.85	33.99	33.22	32.08	29.66	21.63
Cetane No.	50	46	43	36	35	33	32	18	-
Flash Point °C	68	71	73	75	78	81	82	92	
Fire Point °C	71	73	76	79	80	84	85	96	

IV. CONCLUSIONS

Utilizing immaculate soybean oil, it is observed that the diesel motor won't work legitimately because of high kinematic consistency & real contrasts in diesel record and ASTM refining properties. Percentage methyl ester of soybean oil meets the kinematic viscosity and pour point properties according to BIS determination. However properties like ASTM Distillation and diesel index does not meet the detail which is more imperative. In this manner utilizing 100% methyl ester of soybean oil in diesel motor is unrealistic. Blends of 10% and 20% methyl ester of soybean oil with diesel meet the kinematic viscosity, pour point, diesel index and ASTM refining temperatures according to BIS particular. Utilization of 10% blend is the most acceptable whereas 20% and 30% blend produces a satisfactory result. Comparing the other important properties of 10% blend viz. Aniline Point, Diesel Index et al, we observe that the results are strikingly similar to high speed diesel properties. We can conclude that the 10% blend thus obtained is a feasible alternative fuel.

REFERENCES

- Graf J., McKenzie A. M. and Popp M. P., "Hedging break even biodiesel production costs using soybean oil features", Journal of Agribusiness vol. 26, pp. 61 – 75, 2008
- Jain S., Sharma M. P., Renewable and Sustainable Energy Reviews, 14, pp. 667, 2010
- Winayanuwattikun P., Kaewpiboon C., Piriyananon K., Tantong S., Thakernarnkit W., Chulalaksananukul W., Biomass and Bioenergies, 32, pp. 1279, 2008
- Singh S.P., Singh D., Renewable and Sustainable Energies Reviews, 14, pp. 200, 2009
- Ma F., Hanna M. A., "Biodiesel Production: A review", Biosource Technology, vol. 70(1), pp. 1- 15, 1999
- Srivastava A., Prasad R., "Triglycerides-based diesel fuels", Renewable and Sustainable Energy Reviews, vol. 4(2), pp. 111 – 123, 2000
- Mittelbach M., Remschmidt C., "Biodiesel: the comprehensive handbook", M. Mittelbach, Austria.
- Aliyu A., Godwin O. and Hamza A., "Biodiesel production from waste soybean oil", Pelagia Research Library, vol. 2(2), pp. 286 – 289, 2011
- Leung D.Y.C., Wu X., Leung M.K.H., Applied Energy, vol. 87, pp. 1083, 2010
- Knothe G., Van Gerpen J. H. and Krahl J., "The biodiesel handbook", AOCS Press, Champaign, Illinois, 2005
- Gerpen J. V., "Biodiesel Processing and Production", Fuel Processing Technology, vol. 86(10), pp. 1097 – 1107, 2005
- Pahl G., "Biodiesel: growing a new energy economy", Chelsea Green Pub., White River Junction, Vt., 2005
- Saribiyik O. Y., Ozcanli M., Serin H., Serin S. and Aydin K., "Biodiesel Production from Ricinus Communis Oil and its blends with Soybean Biodiesel", Journal of Mechanical Engineering, vol. 56, pp. 811 – 816, 2010
- Hribernik A. and Kegl B., "The influence of biodiesel on combustion and emission characteristics of a diesel engine", Journal of Mechanical Engineering, vol. 53, pp. 683 – 695, 2007
- Zhang Y., Dube M.A., McLean D.D. and Kates M., "Biodiesel Production from cooking oil: 1. Process design and technological assessment", Bioresource Technology, vol. 90, pp. 1 – 16, 2003

International Journal of Innovative Research in Science, Engineering and Technology

(An ISO 3297: 2007 Certified Organization)

Vol. 3, Issue 9, September 2014

16. Martins M. I., Pires R. F., Alves M. J., Hori C. E., Reis M. H. M and Cardoso V. L., "Transesterification of soybean oil for biodiesel production using hydrotalcite as basic catalyst", Chemical Engineering Transactions, vol. 32, pp. 817 – 822, 2013
17. Da Silva N. L., Batistella C.B., Maciel Filho R. and Maciel M. R.W., "Investigations of Biofuel Properties", Chemical Engineering Transactions, vol. 25, pp. 851 – 856, 2011

BIOGRAPHY



Prof. Dr. Ghyas Ahmed Usmani is working as Professor and Head in Department of Oil Technology, University Institute of Chemical Technology, North Maharashtra University, Jalgaon. He has presented and published more than 30 international research papers in journal and conferences. His area of interest includes processing of edible oils, lipase catalyzed interesterification reactions for the production of oleochemicals from non- traditional oils, up-gradation of rice bran oil & recovery of valuable oleochemicals by distillative deacidification and novel surfactants. He has about 23 years of teaching experience and working on various academic and regulatory committees in North Maharashtra University.



Mr. Ajayagiri Goswami is working as Associate Professor in Department of Chemical Engineering, University Institute of Chemical Technology, North Maharashtra University, Jalgaon. He has presented and published more than 15 international research papers in journal and conferences. His area of interest includes biofuel, Mass transfer operations like extraction & adsorption and process control. He has about 16 years of teaching experience and working on various academic and regulatory committees in North Maharashtra University.